

Q1.

A stretched string has a length of 2.00 m and a mass of 3.40 g. A transverse sinusoidal wave is travelling on this string, and is given by  $y(x, t) = 0.030 \sin(0.75x - 126t)$ , where  $x$  and  $y$  are in meters, and  $t$  is in seconds. What is the magnitude of the tension in this string?

- A) 48 N
- B) 60 N
- C) 17 N
- D) 29 N
- E) 35 N

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Q2.

The average power of a sinusoidal wave on a stretched string is  $P$ . If an identical wave is sent simultaneously along the same string in the same direction but with a phase difference of  $90^\circ$  from the first wave, the new average power is

- A)  $2P$
- B)  $P$
- C)  $4P$
- D)  $\sqrt{2}P$
- E)  $P/2$

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Q3.

For a standing wave on a string fixed at both ends

- A) the midpoint is an antinode for odd harmonics.
- B) the midpoint is an antinode for even harmonics.
- C) the midpoint is a node for odd harmonics.
- D) the shortest wavelength corresponds to the fundamental mode.
- E) the amplitude of all points on the string is the same.

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Q4.

A string that is stretched between fixed supports oscillates in a third-harmonic standing wave pattern. The displacement of the wave is given by  $y(x, t) = (0.10) \sin(\pi x/5) \cos(12\pi t)$ , where  $x$  and  $y$  are in meters, and  $t$  is in seconds. What is the length of the string?

- A) 15 m
- B) 6.7 m
- C) 10 m
- D) 60 m

E) 25 m

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Q5.

A string that is stretched between fixed supports has resonant frequencies of 385 and 430 Hz, with no intermediate resonant frequencies. What is the frequency of the seventh harmonic?

- A) 315 Hz
- B) 45 Hz
- C) 2700 Hz
- D) 655 Hz
- E) 3010 Hz

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Q6.

If the intensity of a sound wave traveling in air with constant frequency is doubled, then

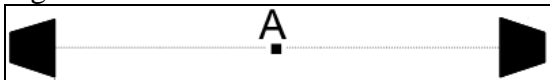
- A) the wave speed remains the same.
- B) the displacement amplitude remains the same.
- C) the displacement amplitude is doubled.
- D) the sound level is doubled.
- E) the displacement amplitude is halved.

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Q7.

Two speakers, separated by 2.00 m, face each other as shown in **Figure 1**. They are driven by the same generator, and emit sound waves with a frequency of 170 Hz, that are initially in phase. A listener is initially at point **A**, which is at the midpoint between the two speakers. What is the shortest distance he should move to find a point of destructive interference? [Take the speed of sound to be 340 m/s]

Fig#



- A) 0.500 m
- B) 0.125 m
- C) 0.250 m
- D) 0.375 m
- E) 0.625 m

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Q8.

A tube open at both ends has length  $L_A$ . A tube open only at one end has length  $L_B$ . If the two tubes have the same fundamental frequency, then

- A)  $L_A = 2L_B$
- B)  $L_A = L_B/2$
- C)  $L_A = L_B/4$
- D)  $L_A = L_B$
- E)  $L_A = 4L_B$

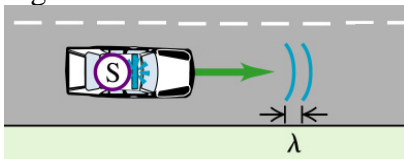
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Q9.

A police car, moving at 20.0 m/s, emits a sound wave with a frequency of 300 Hz. Find the wavelength of the sound wave in front of the car, as shown in **Figure 2**.

[Take the speed of sound in air to be 340 m/s]

Fig#



- A) 1.07 m
- B) 1.13 m
- C) 1.20 m
- D) 0.938 m
- E) 0.833 m

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Q10.

The melting point of sulfur is 444.6 °C and is 586.1 F° below its boiling point. Determine the boiling point of sulfur in degrees Celsius.

- A) 770.2 °C
- B) 118.0 °C
- C) 1031 °C
- D) 1500 °C
- E) 214.2 °C

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Q11.

An iron tank is completely filled with  $2.80 \text{ m}^3$  of water when both the tank and the water are at a temperature of  $32.0 \text{ }^\circ\text{C}$ . When the tank and the water have cooled to  $18.0 \text{ }^\circ\text{C}$ , what additional volume of water can be put into the tank? [ $\alpha_{\text{iron}} = 12.0 \times 10^{-6} / \text{C}^\circ$ ,  $\beta_{\text{water}} = 4.79 \times 10^{-4} / \text{C}^\circ$ ]

- A)  $17.4 \times 10^{-3} \text{ m}^3$
- B)  $1.41 \times 10^{-3} \text{ m}^3$
- C)  $18.8 \times 10^{-3} \text{ m}^3$
- D)  $0.470 \times 10^{-3} \text{ m}^3$
- E)  $18.3 \times 10^{-3} \text{ m}^3$

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Q12.

A 100-g ice cube at  $0.0 \text{ }^\circ\text{C}$  is placed in 650 g of water at  $18 \text{ }^\circ\text{C}$ . If the system is isolated, what is the final temperature?

- A)  $5.0 \text{ }^\circ\text{C}$
- B)  $0.0 \text{ }^\circ\text{C}$
- C)  $22 \text{ }^\circ\text{C}$
- D)  $28 \text{ }^\circ\text{C}$
- E)  $12 \text{ }^\circ\text{C}$

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Q13.

A copper rod has a length of 60 cm. One end is maintained at  $80 \text{ }^\circ\text{C}$  and the other end is at  $20 \text{ }^\circ\text{C}$ . In steady state, what is the temperature of the rod at a point which is 20 cm from the hot end?

[ $k_{\text{copper}} = 401 \text{ W/m.K}$ ]

- A)  $60 \text{ }^\circ\text{C}$
- B)  $40 \text{ }^\circ\text{C}$
- C)  $35 \text{ }^\circ\text{C}$
- D)  $25 \text{ }^\circ\text{C}$
- E)  $50 \text{ }^\circ\text{C}$

Q14.

A 5 moles of an ideal gas expand isobarically from  $T_i = 25\text{ }^\circ\text{C}$  to  $T_f = 75\text{ }^\circ\text{C}$ . Calculate the work done by the gas during this process.

- A)  $2.1 \times 10^3\text{ J}$
- B)  $4.5 \times 10^3\text{ J}$
- C)  $1.2 \times 10^3\text{ J}$
- D)  $5.4 \times 10^3\text{ J}$
- E) zero

Q15.

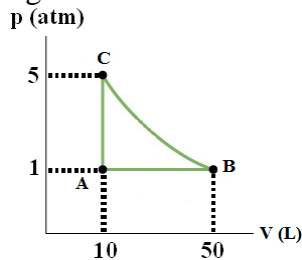
An ideal gas has a density of  $3.75\text{ kg/m}^3$  and is at a pressure of  $1.00\text{ atm}$ . Determine the rms speed of the molecules of this gas.

- A)  $284\text{ m/s}$
- B)  $94.9\text{ m/s}$
- C)  $164\text{ m/s}$
- D)  $1070\text{ m/s}$
- E)  $616\text{ m/s}$

Q16.

An ideal monatomic gas is taken through cycle  $A \rightarrow B \rightarrow C \rightarrow A$ , shown in the  $p$ - $V$  diagram of **Figure 3**, where process  $B \rightarrow C$  is isothermal. Calculate the net work done in one cycle.

Fig#



- A)  $4088\text{ J}$ , on the gas
- B)  $4088\text{ J}$ , by the gas
- C)  $8152\text{ J}$ , by the gas
- D)  $8152\text{ J}$ , on the gas
- E)  $4000\text{ J}$ , by the gas

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Q17.

One mole of an ideal monatomic gas is initially at a pressure of  $1.01 \times 10^5$  Pa, a temperature of 300 K, and has a volume of 1.00 L. It is compressed adiabatically to a volume of 0.0667 L. Calculate the magnitude of the work done during this process.

- A) 19.0 kJ
- B) 6.75 kJ
- C) 94.3 kJ
- D) 2.49 kJ
- E) Zero

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Q18.

A system consists of two large thermal reservoirs in contact with each other, one at a temperature of 300 °C and the other at a temperature 200 °C. If 600 J of heat is transferred from the 300 °C reservoir to the 200 °C reservoir, what is the change in entropy of this system?

- A) 0.221 J/K
- B) 1.00 J/K
- C) 5.00 J/K
- D) -1.00 J/K
- E) 2.31 J/K

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Q19.

A Carnot refrigerator is operated between two heat reservoirs at temperatures of 320 K and 270 K. In each cycle, the refrigerator extracts 415 J of heat from the cold reservoir. If the refrigerator completes 165 cycles each minute, what is the power input required to operate it?

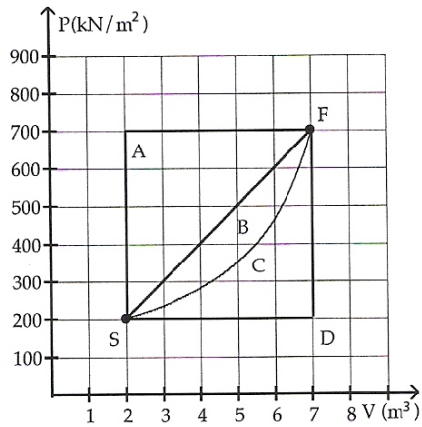
- A) 211 W
- B) 178 W
- C) 815 W
- D) 224 W
- E) 317 W

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Q20.

Which of the processes on an ideal gas shown in **Figure 4** results in the minimum change in entropy of the gas in changing the gas from state S to State F?

Fig#



- A) All processes
  - B) Only process A
  - C) Only process B
  - D) Only process C
  - E) Only process D
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